EXAFS data analysis of supported metal catalysts and Fuel Cell electrocatalysts

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•EXAFS: introduction to data analysis

 Example. Dopants and defects: local structure in barium cerates and zirconates

•On line tutorial (VIPER package)

THE EXAFS SPECTRUM



Data analysis: get quantitative information

XAFS ANALYSIS: from the experimental data to results



DATA Collection

- a Choose properly the experimental set-up (XANES/EXAFS Fluo/TEY/Transmission) and Sample preparation
- b Check the data quality During the experiment

Quality is better than the quantity

c Choose properly the data collection strategy

d

Always measure Reference data

XAFS rarely gives absolute quantitative details

OPTIMIZE YOUR BEAMTIME

Plan your EXPERIMENT!

Proposal submission + proposal evaluation + beam time scheduling = <u>6 +12 months</u>! DUBBLE: 1-10 + 1-04 ESRF: 01-09 + 01-03

It is difficult to obtain new beam time for the same proposal In case of failure

-Check the proposal deadline -discuss your experiment with local contacts -check your data during the experiment

DATA Collection

Mistake in XAFS data analysis are dominated by systematic errors:

Moreover:

- spurious structures
- background subtraction
- wrong models
- inaccurate theory

Experimental set-up & sample preparation

 Transmission: massive concentrated samples inhomogeneities, holes, not parallel surfaces, etc...
 Fluorescence: thin concentrated thick diluted samples

Self absorption, detector linearity, Bragg reflections 3. TEY: thick concentrated

Surface sensitivity, sample charge, Bragg reflections

Data quality

Noise evaluation glitches discontinuities

Collection strategy

- step scan / quick EXAFS
- Fixed Energy or K steps
- single scan/repeated scans
- temperature points ...

Measure reference samples

1-energy calibration 2-data analysis procedures calibration

The power of XAFS technique consists in highlighting the smallest structural differences

NOISE evaluation



 $S/N \sim J/s$

It must be Checked regularly during the experiment!

Check systematically during the experiment



XAFS ANALYSIS: from the experimental data to results



Preliminary data treatment

Select properly the best spectra and useful data regions

Preliminary data treatment

- Self absorption corrections
- de-glitching (if it possible!)
- data interpolation
- data alignment

Sum together the best spectra

Weighting by S/N



XAFS ANALYSIS: from the experimental data to results



Extract $\chi(k)$ structural signal



EXAFS analysis: step 2



 μ_0 = represents the bare atom absorption.

It is calculated empirically as a smooth curve across the data. Different XAFS data analysis software apply different (generally equivalent) methods.

Requirements for μ_0 :

 Smooth enough to not remove true structural features
 Structured enough to remove background structures

How to control he extraction?

Large |FT| contributions at low (unphysical) distances may signify "wrong μ_0 "







MAIN EXAFS approximations:



Multi-electron excitations

The approximations outlined neglect several important many-body effects.

A first class of effects, usually referred to as '*intrinsic*', is that associated with the multi-electron response of the photoabsorber atom.

besides the core level involved in the main edge, **non-negligible transition** are excited

All these transitions contribute to a complex atomic background shape.

The intensity of these channels is predicted to be of the order of up to a few % of the main Kedge channel.

Thanks to Adriano Filipponi J. Phys.: Condens. Matter 13 (2001) R23-R60





A. Di Cicco et Al. Phys. Rev. B 62, 12001 (2000)

Multiple excitations can affect the background with (small) discontinuities visible as peaks at distances smaller than 1



Figure 5. The atomic background of the K edge of Zr and Br, and the L_3 edge of Hg in Zr foil, CuBr, and HgO, respectively.

J. Phys.: Condens. Matter 9 (1997) 5507–5515.



XAFS ANALYSIS: from the experimental data to results





EXAFS analysis: Fourier Filtering



Peak positions are not the true distances due to the photoelectron phase shift!

Shape of FT widely changes as a function of:

- FT window
- kw weight
- data range

FT shows more intuitively the main structural features in the real space: the FT modulus represent a *pseudoradial distribution function* modified by the effect of amplitude, phase and mean free path parameters: |FT| peaks point out interatomic correlations

EXAFS analysis: Fourier Filtering



 N_i number of coordination ith shell, R_i ith distance between the absorber and ith shell σ_i is the Debye-Waller factor (disorder term) of the ith shell

EXAFS analysis: Fourier Filtering thermal factor



Data (structural) refinement



HOW choose a structural model



XAFS data analysis software



EXAMPLE ANALYSIS



WebATOMS version 1.8 (Atoms 3.0beta10) (3 February, 2005) ATOMS is copyright © 1998-2005 Bruce Ravel

> Visit ATOMS and FEFF webpages for more information!



Sh R	Ν	RAu(Å)
I: a/sqrt(2)	12	2.880
II: a	6	4.0789
III: axsqrt(1.5)	24	
IV: axsqrt(2)	12	
V: axsqrt(2.5)	24	

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EXCHANGE 0	Keep limit= 0.00 Heap limit 0.00 Pwcrit= 2.50%	
<pre>*** I_scf = 0 for a solid, 1 for a molecule * r_scf [I_scf n_scf ca]</pre>		
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*** L fms = 0 for a solid 1 for a molecule	feff0004.dat 0.00000 5.109 48.000 3 4.9250	
* r_fms l_fms	feff0005.dat 0.00000 56.390 24.000 2 4.9970	
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Fitting procedure





Visualize the structure and understand the neighbor shells!



... and take care about multiple scattering contributions



Important when the angle is > 120° . The MS at 180° has highest contribution due to the "focusing effect"

How to choose the MS path

ATH Rmax= 7.100, Keep limit= 0.00, Heap limit 0.00 Pwcrit= 2.50% 1 2 12.000 index, nleg, degeneracy, r= 2.8850 z ipot label rleq beta eta 2.8850 180.0000 0.0000 0.000000 -2.040000 2.040000 1 'Au 0.000000 0.000000 0.000000 0 'Au 2.8850 180.0000 0.0000 2 2 6.000 index, nleg, degeneracy, r= 4.0800 z ipot label rleq beta V eta х -4.080000 0.000000 0.000000 1 'Au 4.0800 180.0000 0.0000 4.0800 180.0000 0.0000 0.000000 0.000000 0.000000 0 'Au 3 3 48.000 index, nleg, degeneracy, r= 4.3275 z ipot label rlea beta eta Х V 2.040000 2.040000 0.000000 1 'Au 2.8850 120.0000 0.0000 0.000000 2.040000 -2.040000 1 'Au 2.8850 120.0000 0.0000 0.000000 0.000000 0.000000 0 'Au 2.8850 120.0000 0.0000 4 3 48.000 index, nleg, degeneracy, r= 4.9250 z ipot label rleq beta eta х V 4.080000 0.000000 0.000000 1 'Au 4.0800 135.0000 0.0000 2.040000 2.040000 0.000000 1 'Au 2.8850 90.0000 0.0000 0.000000 0.000000 0.000000 0 'Au 2.8850 135.0000 0.0000 5 2 24.000 index, nleg, degeneracy, r= 4.9970 z ipot label rlea beta eta Х V -2.040000 -2.040000 4.080000 1 'Au 4.9970 180.0000 0.0000 0.000000 0.000000 0.000000 0 'Au 4.9970 180.0000 0.0000 6 3 48.000 index, nleg, degeneracy, r= 5.3835 ipot label Z rleq beta eta Х 2.8850 150.0000 -2.040000 -2.040000 0.000000 1 'Au 0.0000 2.040000 0.000000 2.040000 1 'Au 4.9970 150.0000 0.0000 0.000000 0.000000 0.000000 0 'Au 2.8850 60.0000 0.0000 7 3 96.000 index, nleg, degeneracy, r= 5.3835 ipot label rleg beta Z eta Х 4.080000 -2.040000 -2.040000 1 'Au 4.9970 150.0000 0.0000 2.040000 -2.040000 0.000000 1 'Au 2.8850 60.0000 0.0000 0.000000 0.000000 0.000000 0 'Au 2.8850 150.0000 0.0000 8 2 12.000 index, nleg, degeneracy, r= 5.7700 Z ipot label rleq beta eta х -4.080000 0.000000 4.080000 1 'Au 5.7700 180.0000 0.0000 0.000000 0.000000 0.000000 0 'Au ' 5.7700 180.0000 0.0000 9 3 12.000 index, nleg, degeneracy, r= 5.7700







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Dopants and defects: local structure in barium cerates and zirconates

Proton conductors



Electrolyte: Y:BaZr(Ce)O₄

Anode: Ni/(YSZ)

CERMET

Anionic conductors





Anode : $H_2 + O^{2-} = H_2O + 2e^{-}$ Problems:

High temperature Chemical stability

- . Low ion/electron conductivity
- . High production cost

Open questions: conduction mechanisms, its correlation with the structure and perovskite chemistry

ABO₃ perovskite (e.g. BaCeO₃ = BCY)

B-site smaller Ce⁴⁺ cation



Doping $2 \text{ Ce}_{\text{Ce}}^{x} + O_{0}^{x} + Y_{2}O_{3}$ $= 2 \text{ YC}_{e}^{'} + V_{0}^{*} + 2 \text{ CeO}_{2}^{'}$

A-site larger Ba²⁺ cation

(De)-hydration $H_2O(g) + V_0^{*} + O_0^{*} = 2 OH_0^{*}$



Averaging: higher thermal factors, modified positions deceiving!

Takeuchi et al, SSI 2000

Azad et al, J Mater Chem 2008

But <u>only</u> neutron scattering/XRD, i.e. no local information

Local: two different local environments, each with its own features

Lots of simulations but almost <u>no measurements</u> until 2005

Local information required

Diffraction can be deceiving, because it averages out doped and undoped sites that are not equivalent from a local point of view!

EXAFS is a good complementary technique

In situ EXAFS of Y:BaCeO₃



EXAFS collected @ BM26 and BM08

Longo et al. Chem. Mater. (2006).



Figure 5. Fourier transform of XAFS data and model relative to dry $BaCe_{0.98}Y_{0.02}O_{3-\delta}$. The model FT corresponds to a regular octahedral oxygen shell around yttrium.



Figure 8. Fourier transform of XAFS data and model of the dry $BaCe_{0.98}Y_{0.02}O_{3-\delta}$ assuming a distorted octahedral yttrium environment.

Y³⁺ in BCY



550°C in wet flux



Longo et al. Chem. Mater. (2006).





FT magnitude (a.u.)

Giannici, Longo et al Solid State Ionics (2007).

Gd³⁺ in BCG

O octahedron expands

Proton (de)insertion has strong effects on dopant environment



Ba polyhedron contracts (!?)



EXAFS @ BM29

Giannici et al. Chem. Mater. (2009).

F. Giannici et al. Solid State Ionics (2010).





mobility

Ga < Sc < In < Gd < Y

disorder

EXAFS @ BM29

Giannici , Longo et al. Chem. Mater. (2011). ESRF Highlights (2011) EXAFS: introduction to data analysis

 example: Dopants and defects: local structure in barium cerates and zirconates

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THANK you for your attention